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Name _____

Physical and Chemical Changes in Matter

Write Physical Change or Chemical Change for each example below.

An ice cube melts _____	Rust forms on a car _____
A glass bottle breaks _____	A sheet of paper is cut in half _____
An apple is cut _____	A log burns at a campfire _____
A piece of wood is painted _____	Cooking an egg _____

LIVEWORKSHEETS

Name _____ Date _____ Class _____

Directed Reading for **Overview** Content Mastery **Elements and Their Properties**

Directions: Use the terms below to correctly complete the statements. Write the terms in the blanks to the left. Then find and circle each term in the puzzle.

- | | | | |
|------------|------------|-------------|------------|
| metals | lanthanide | alkali | salts |
| metalloids | hydrogen | noble gases | allotropes |

- _____ 1. Elements with atomic numbers from 58 through 71 are part of the _____ series.
- _____ 2. _____ have both metallic and nonmetallic properties.
- _____ 3. Diamond and graphite are _____ of carbon.
- _____ 4. All of the elements in the boron group except boron are _____.
- _____ 5. Halogens form _____ when combined with metals.
- _____ 6. The _____ do not naturally mix with other elements to form compounds.
- _____ 7. The _____ metals are the most reactive group of elements.
- _____ 8. _____ is the only nonmetal on the left side of the periodic table.

Meeting Individual Needs

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 U H L O H O A N G R S C O N C E V R A
 R H E C Y Z R P H I G A R J R T N H A
 S E L P D O O B S A L T S O I A M N F
 E P T I R N B S L L C A B I T L U F L
 K M P G O E P M S K A L E N I L A R T
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 S S U S E H A T E L M T E M A I R M C
 G O L A N T H A N I D E C S L D S B A
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